**AMENDMENTS TO THE SPECIFICATION:**

Please replace the paragraph beginning on page 2, line 1 and ending on page 2, line 14, with the following amended paragraph:

As for the above methods, the method (1) using 1,2,4-triazole is unsuitable for mass production, since the reaction for the synthesis of 4-(1,2,4-triazol-1-yl)pyrimidin-2-(1H)-one derivatives is extremely slow and time-consuming, and the extraction of the product is required, thereby taking a lot of time and effort; and the method (2) via 4-[4-(dimethylamino)pyridinium] derivatives is also unsuitable for mass production, since the reaction is extremely slow and time- consuming, and, if less than 2.0 equivalents of DMAP is used based on the reaction substrate, then the unchanged substrates remain and the manipulation to separate them from the product ~~become~~ becomes necessary, resulting in an increase in the number of processes, as described in the comparative examples later.

Please replace the paragraph beginning on page 11, line 11 and ending on page 11, line 24, with the following amended paragraph:

R4 and R5 in ammonia, or a primary or secondary amine represented by formula (2) independently represent a hydrogen atom; an alkyl group having 1 to 4 carbon atoms, i.e. a methyl group, an ethyl group, a propyl group, a 2-propyl group, and a tert-butyl group; a cycloalkyl group having 5 to 8 carbon atoms, i.e. a cyclopentyl group, a cyclohexyl group, a cycloheptyl group, and a cyclooctyl group; an alkyl group having 1 to 4 carbon atoms ~~substitutes~~ substituted with a halogen atom(s), i.e. a chloromethyl group

and a dichloromethyl group; an alkenyl group having 2 to 4 carbon atoms, such as a bromovinyl group, etc. Additionally, R4 and R5 may be linked together to form a ring, and in this case, R4(R5)N group may be, but not limited to, a pyrrolidine group or a piperidine group. Particular preference is given to ammonia and a piperidine.

Please replace the paragraph beginning on page 13, line 8 and ending on page 13, line 16, with the following amended paragraph:

When the acidity within the reaction system ~~affect~~ affects the reaction, in addition to the tertiary amine, a deacidifying agent can be added. In addition to the above tertiary amine, examples of the deacidifying agent include, but are not limited to, organic bases, such as pyridine, lutidine, N,N-dimethylaniline, or inorganic bases, such as potassium carbonate, sodium carbonate, and sodium hydrogencarbonate; and ion-exchange resin. Particular preference is given to triethylamine.

Please replace the title on page 25, line 1, with the following amended title:

~~CLAIMS~~ We claim: